

CO₂ VALUE OF BIO ETHERS

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**Central and Eastern European
Refining and Petrochemicals
13th Annual Meeting**

Budapest, Hungary
12th October 2010



European Fuel Oxygenates Association

WWW.EFOA.ORG



- Non-profit Technical Organisation
- Founded in 1985
- ~ 2/3rd of Total EU Etherification Capacity

CO₂: High Profile in the EU



Overall EU GHGs Emission Reduction Targets

Climate and Energy Package

2020 = 20% *(if EU Unilaterally)*

2020 = 30% *(if Global Agreement)*

Biofuels & Bioliquids' GHGs LC Emission Saving Thresholds

Directive 2009/28/EC "Renewable Energy"

2011 ≥ 35 %

2017 ≥ 50 %

2018 ≥ 60 % *(Plants ≥ 1/2/2017)*

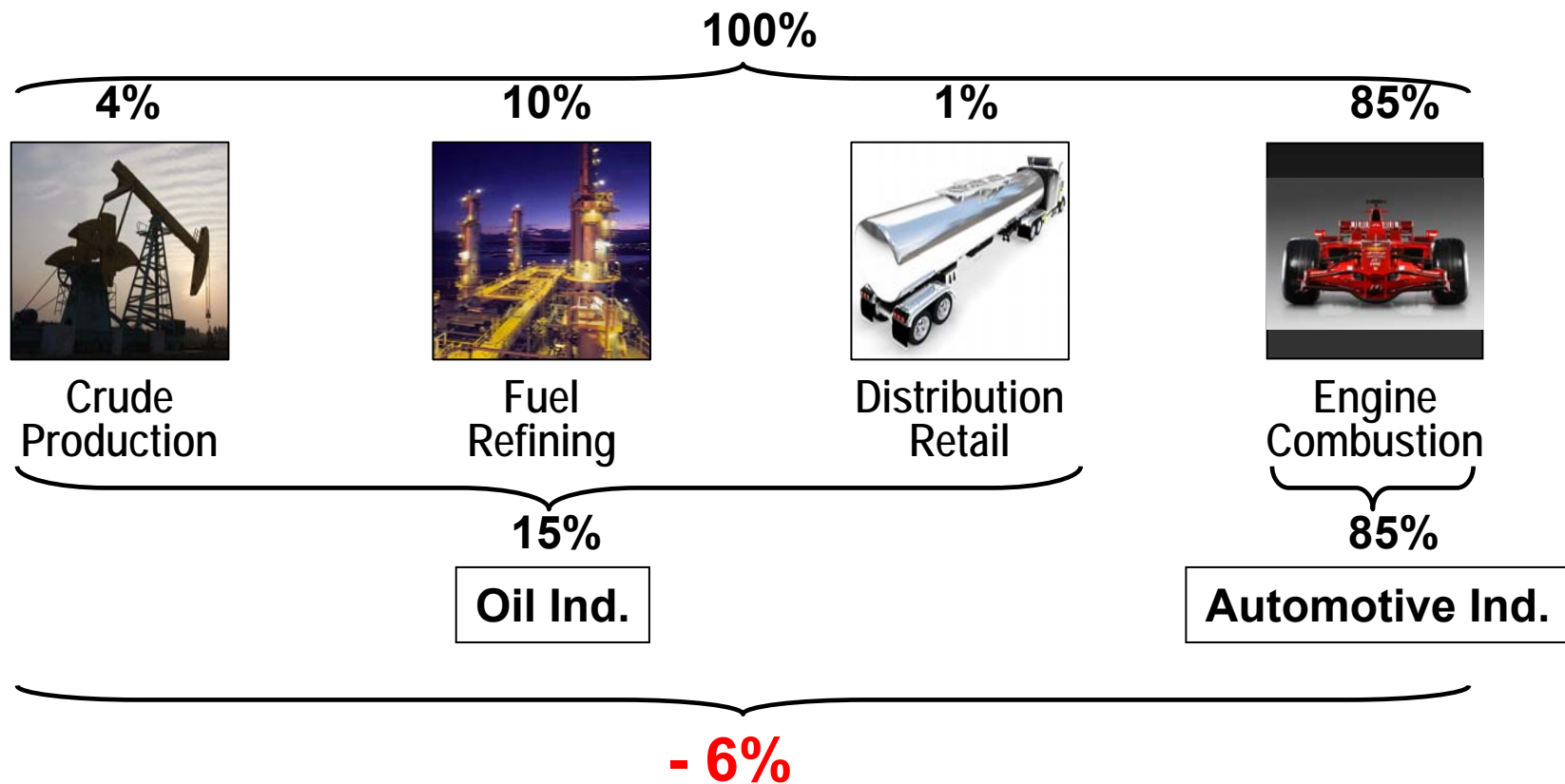
Fuels GHGs Emission Reduction Targets *(per unit of energy)*

Directive 2009/30/EC "Fuel Quality"

2020 = 10 % *(Indicative)*

2020 ≥ 6 % *(Mandatory)*

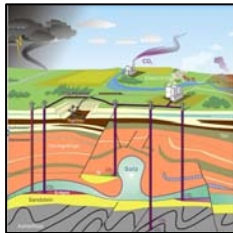
GHGs Emissions Reduction & Refining: - 6% a big Challenge !



Only **Few Options** Available to Refiners:
Bio-fuels key !



Operations Energy Efficiency

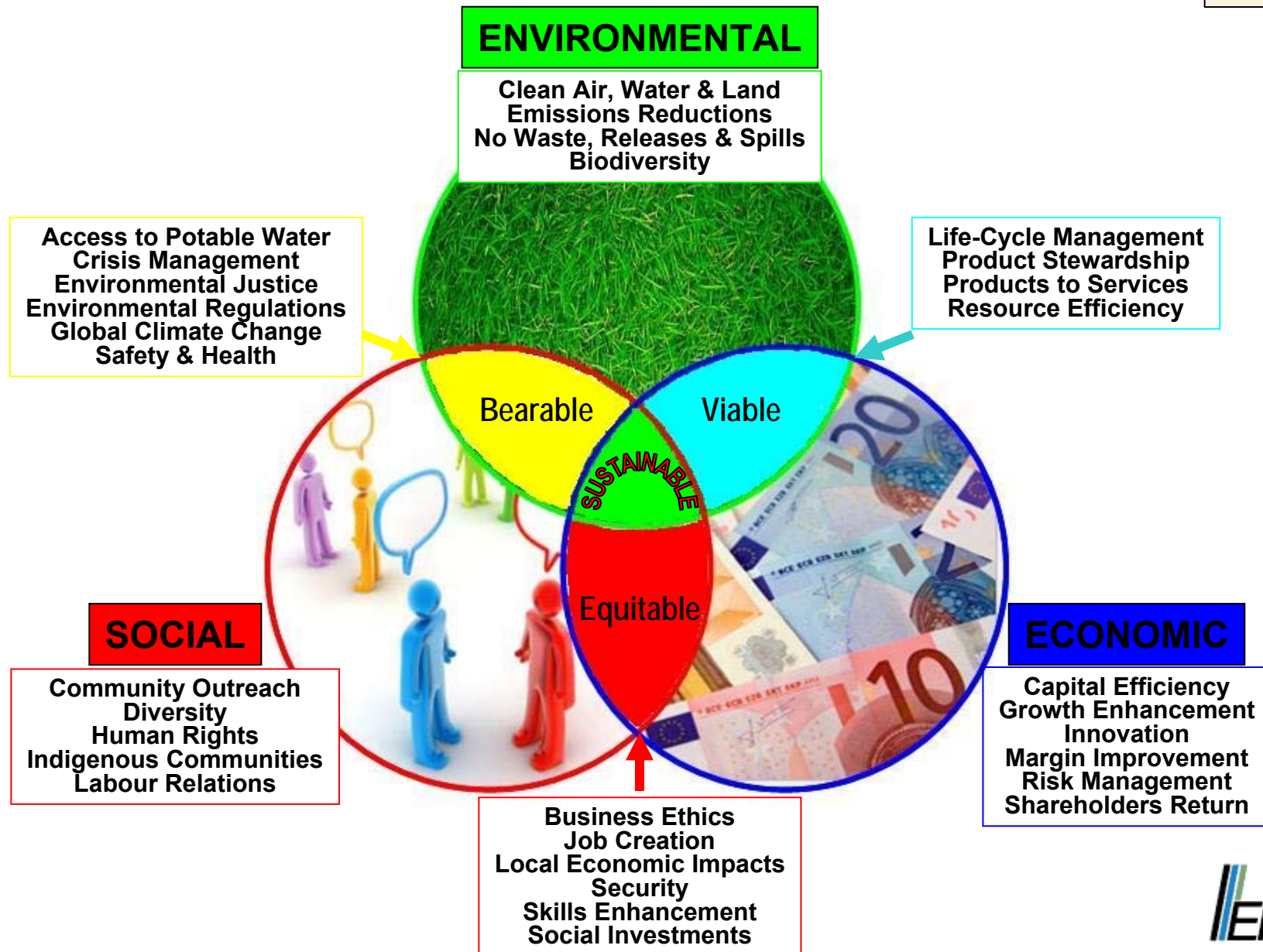


Carbon Capturing and Storage

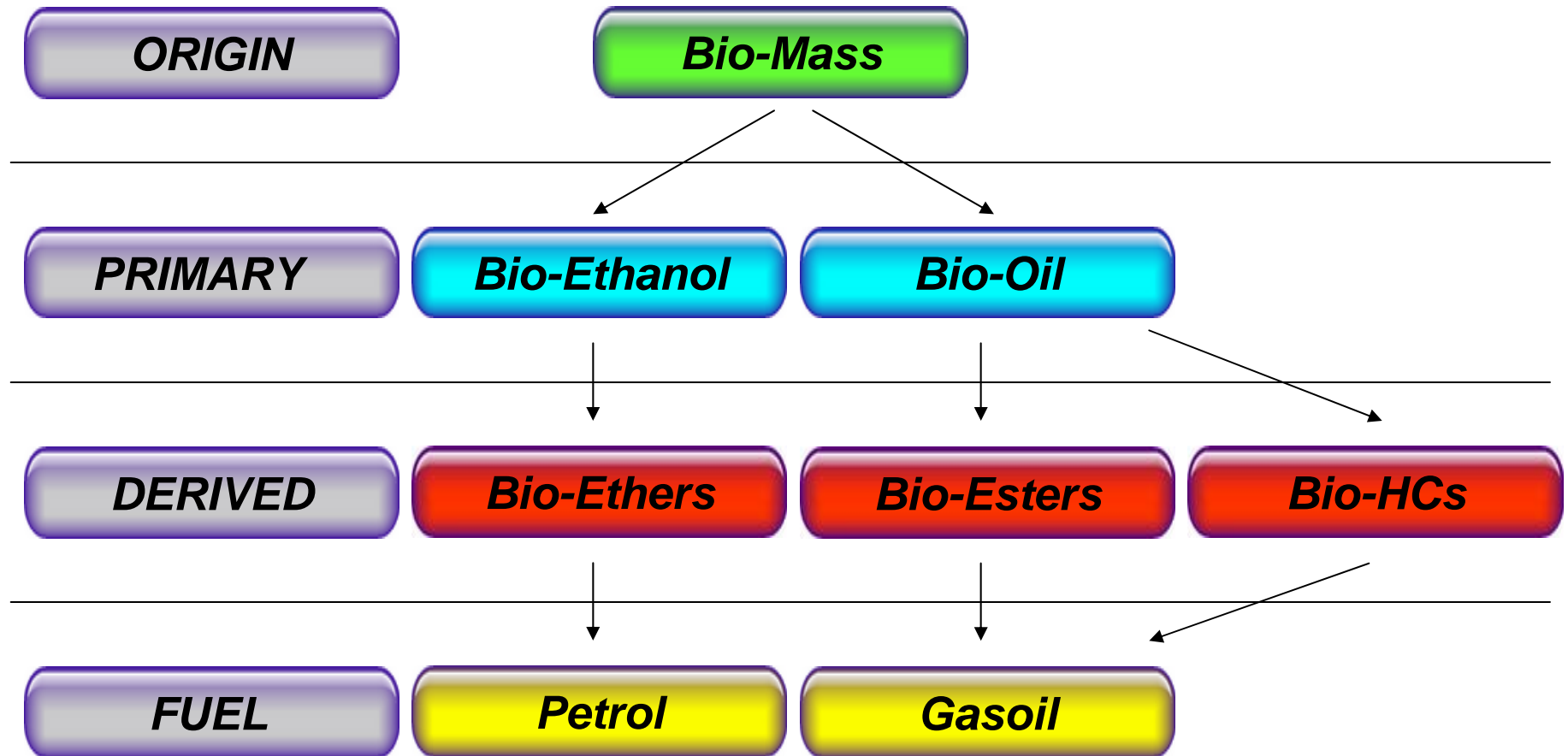


Bio-Fuels

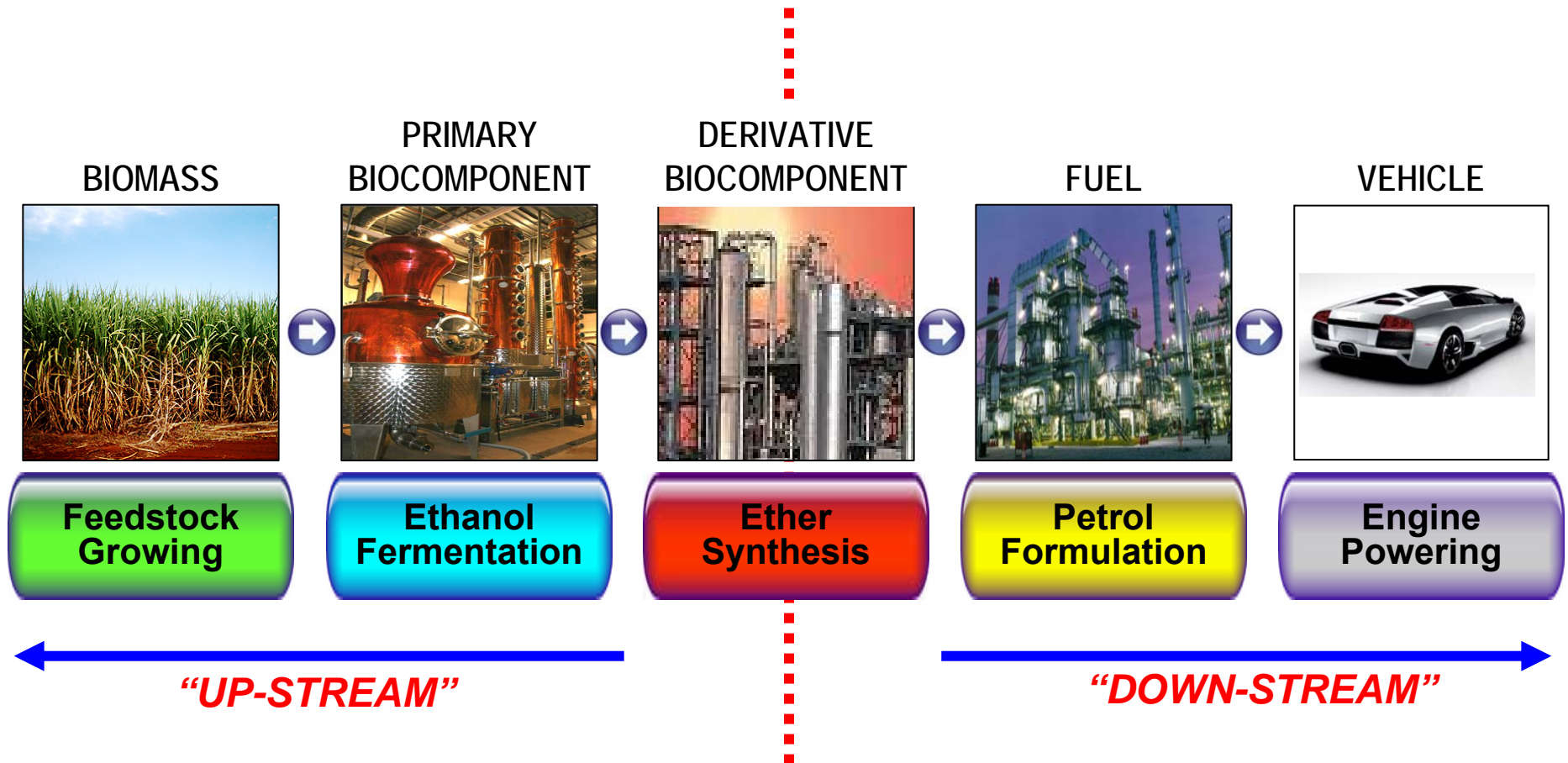
Bio-component to Address ≠ Sustainability Objectives



Bio-Ether (**ETBE**) is for **Petrol**
what Bio-Ester (**FAME**) is for **Gasoil**



Bio-Petrol Supply Chain: **Ethers** in **Central** Position



Bio-Ethers' “Up-stream”: Embedding Ethanol “Goodness”



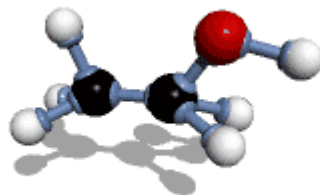
Ethanol

*{ Sustainable, Certified, Audited, ..
1st Generation, 2nd Gen., 3rd Gen...
Cellulosic, from Waste, }*



ETBE

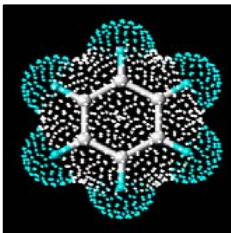
Bio-Ethers' “**Down-stream**”: Enabling Lesser CO₂-Intensive Petrol Formulation



Bio-Ethers Reduce Refinery Operations' CO₂ Emissions



< Carbon Content

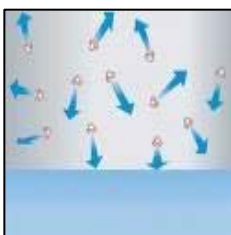
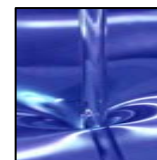


< Aromatics

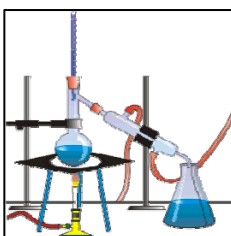


< Refinery Fuel

Key Blending Properties Affecting Fuel Formulation



Vapour pressure

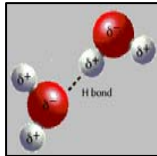


Distillation characteristics



Octane performance

ETBE: Further Enhancing ETOH Performances



← < Blending **Volatility**

< VOCs **Emissions**



← > **Octane** Barrel Delivery

> **CO₂** Emissions Reduction



← > **Crude-oil** Replacement

No **Commingling**



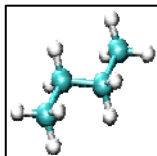
← > **Water** Tolerance

No **Azeotrope**



← < **Logistic** Complexity

> **Blend-stocks** Value

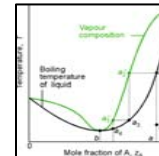


← **Butane** Uptake/Upgrade

> Refining **Flexibility**

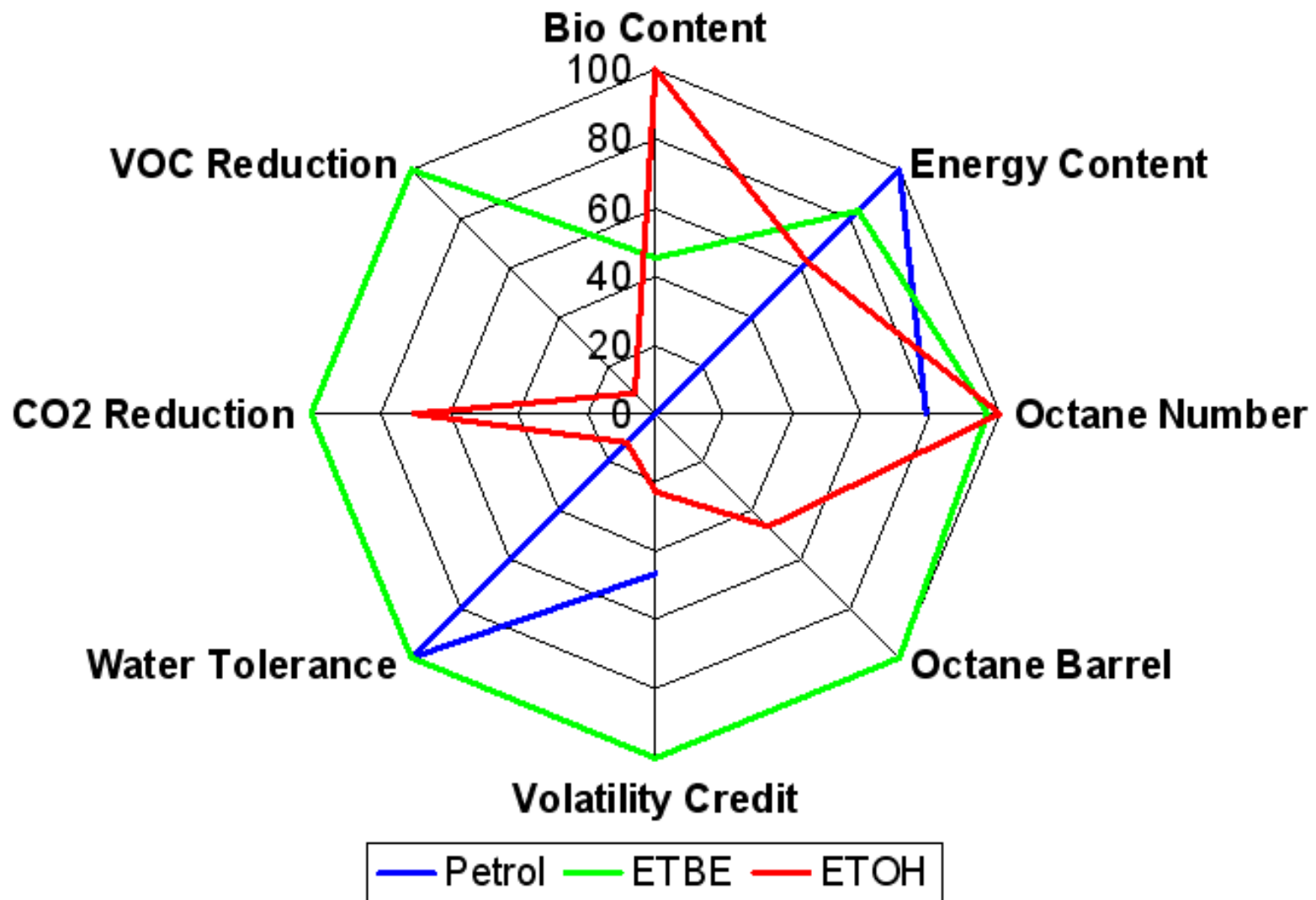


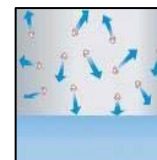
← > Material **Compatibility**



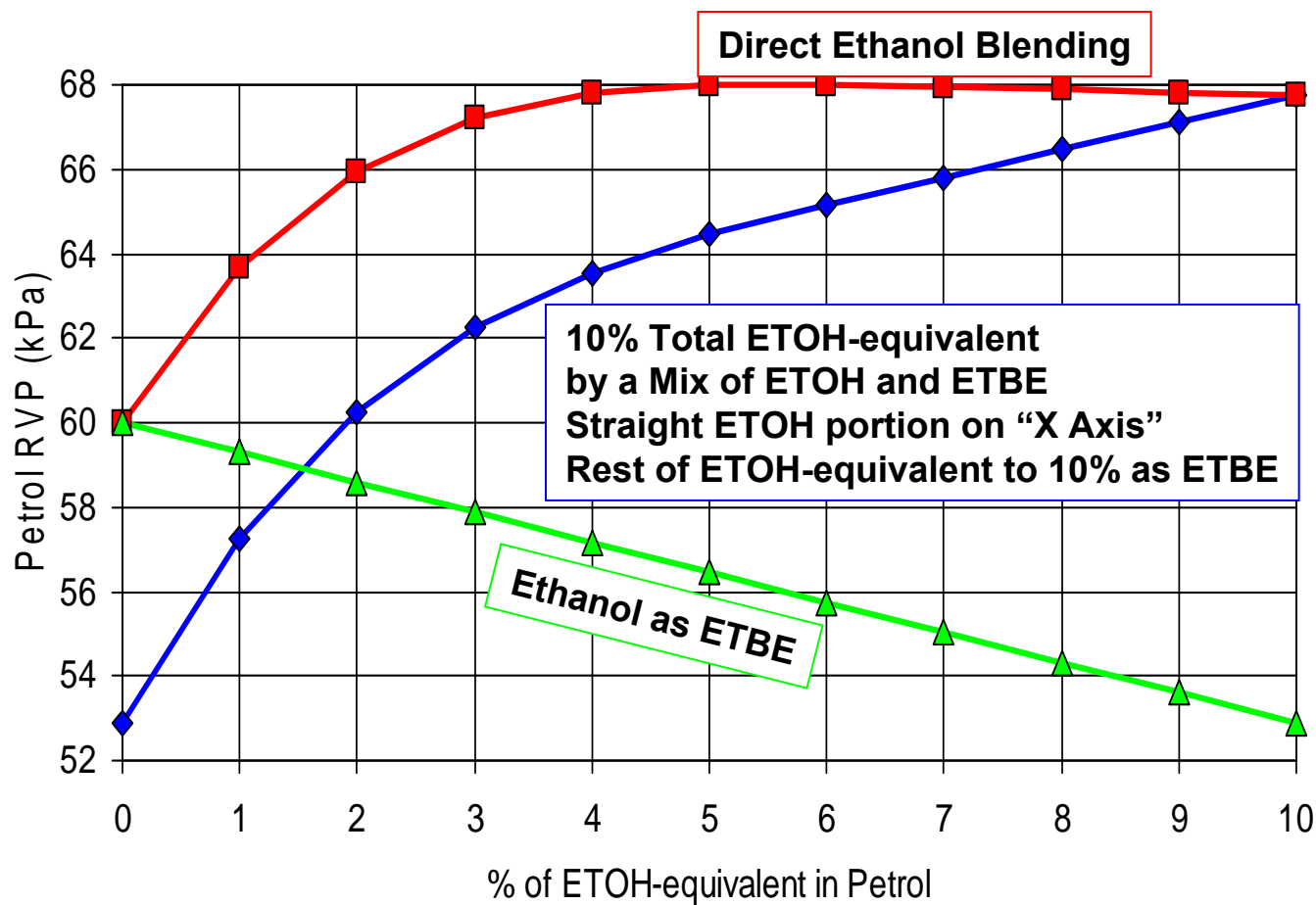
Ethanol/ETBE/Petrol **Blending Performances**

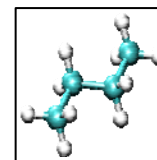
"Normalized" Comparison



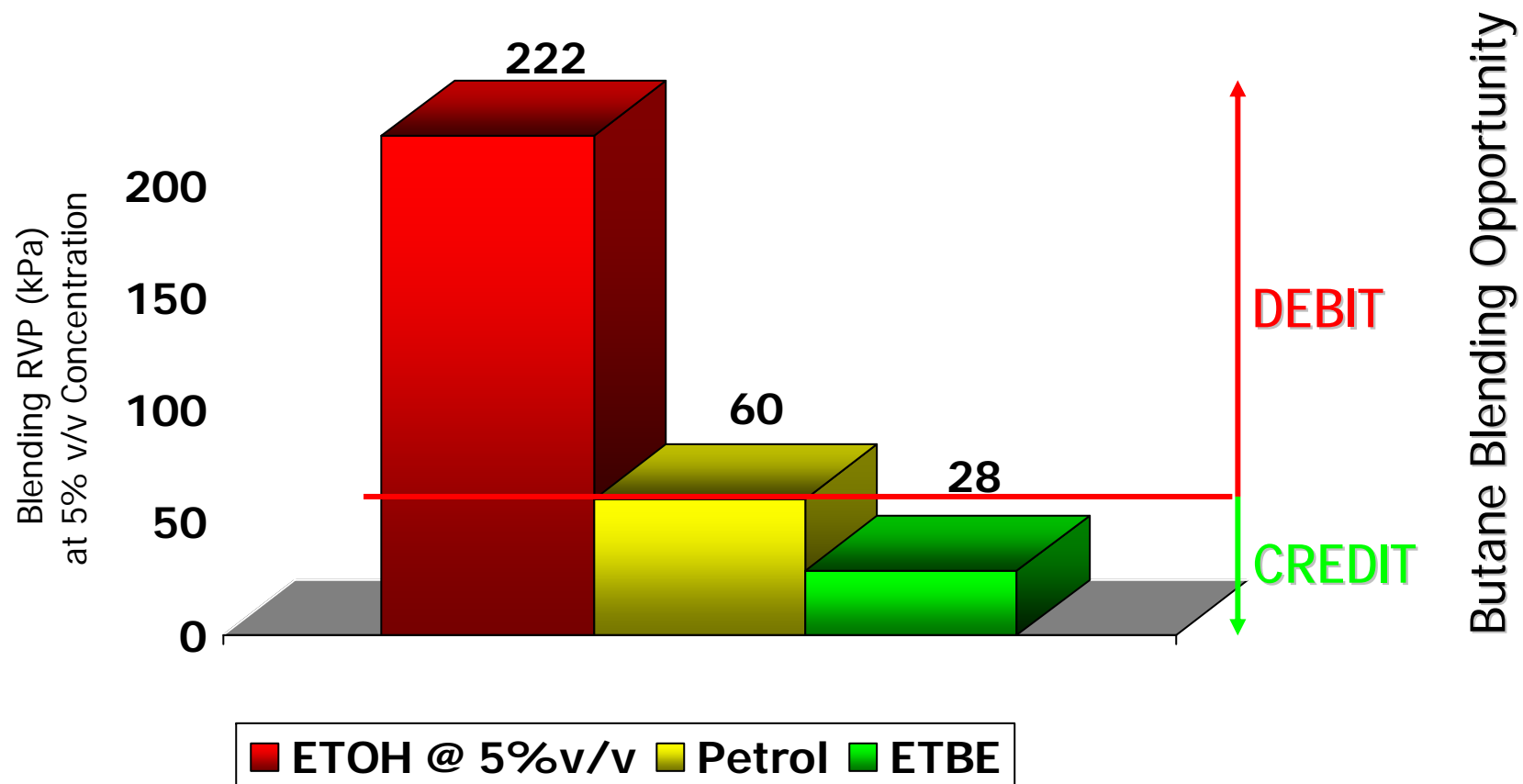


ETBE Helping Ethanol on Volatility

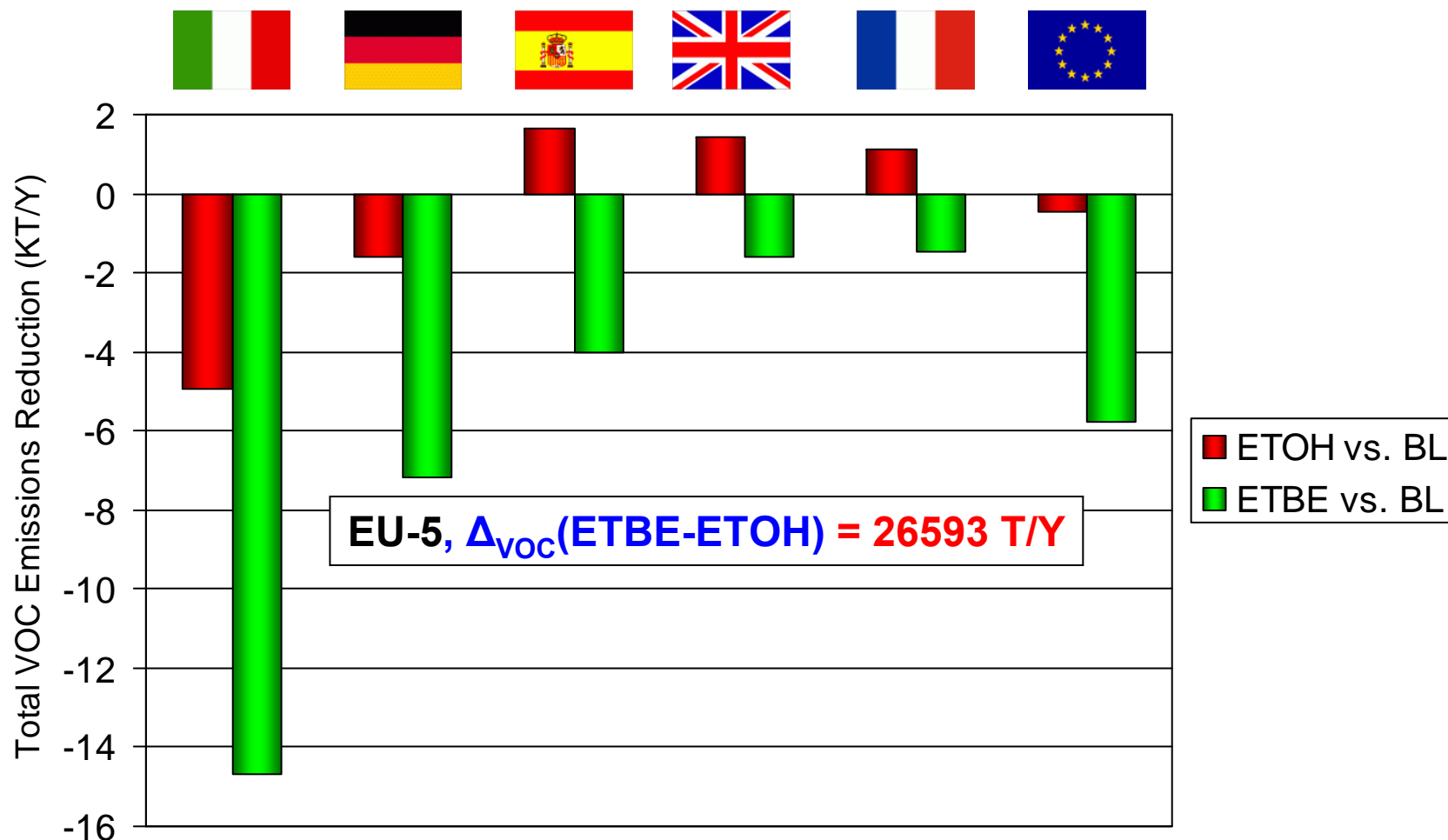




Butane Uptake/Upgrade Value



ETBE Further Addresses GHG's by Greatly Reducing VOCs Emissions

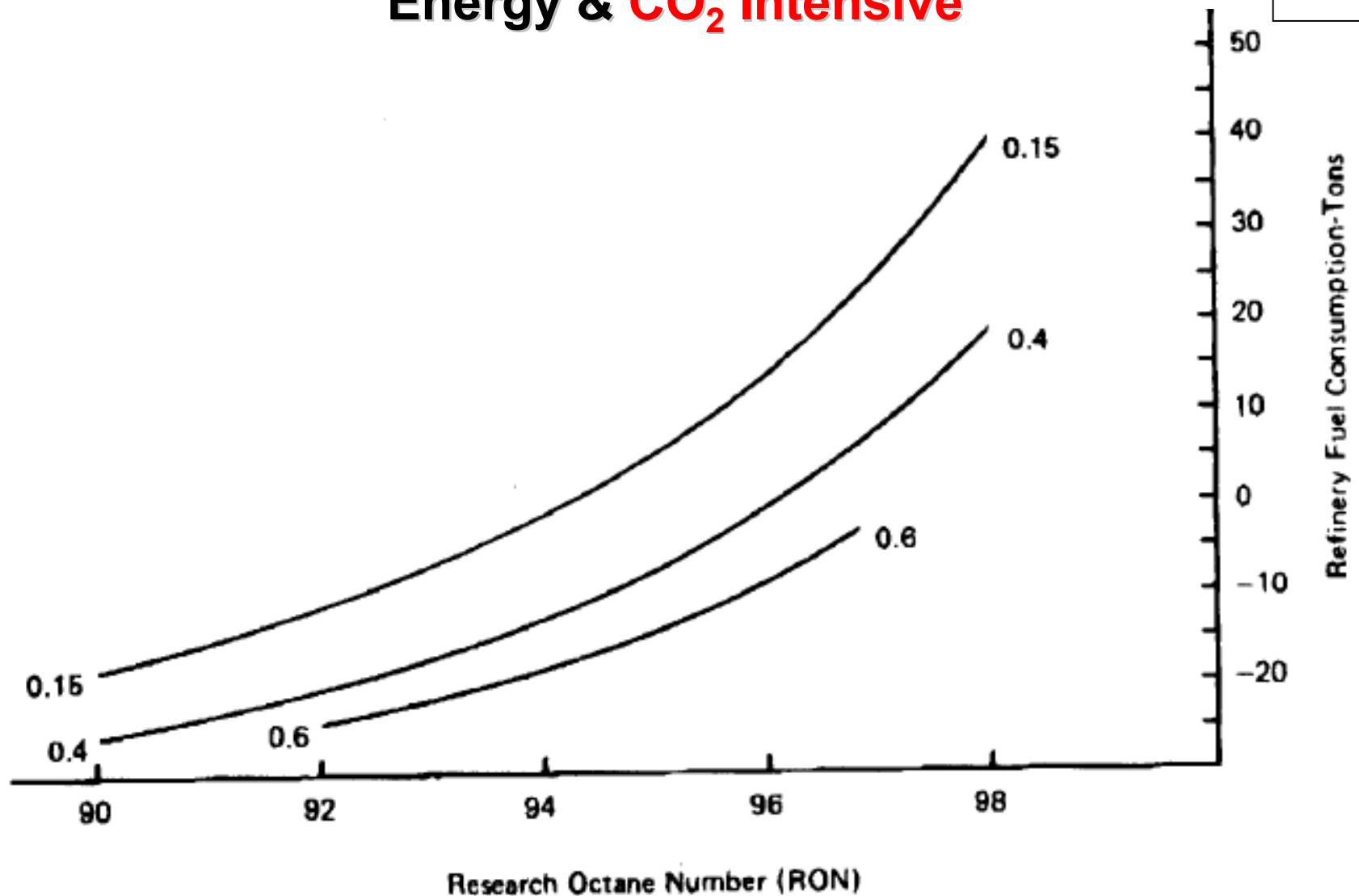


Source: Emissions and Health Unit - Institute of Environment and Sustainability - EC-JRC Ispra

"An assessment of the impact of ethanol-blended petrol on the total NMVOC emissions from road transport in selected countries"

Refinery Octane Production: Energy & CO₂ Intensive

concaawe

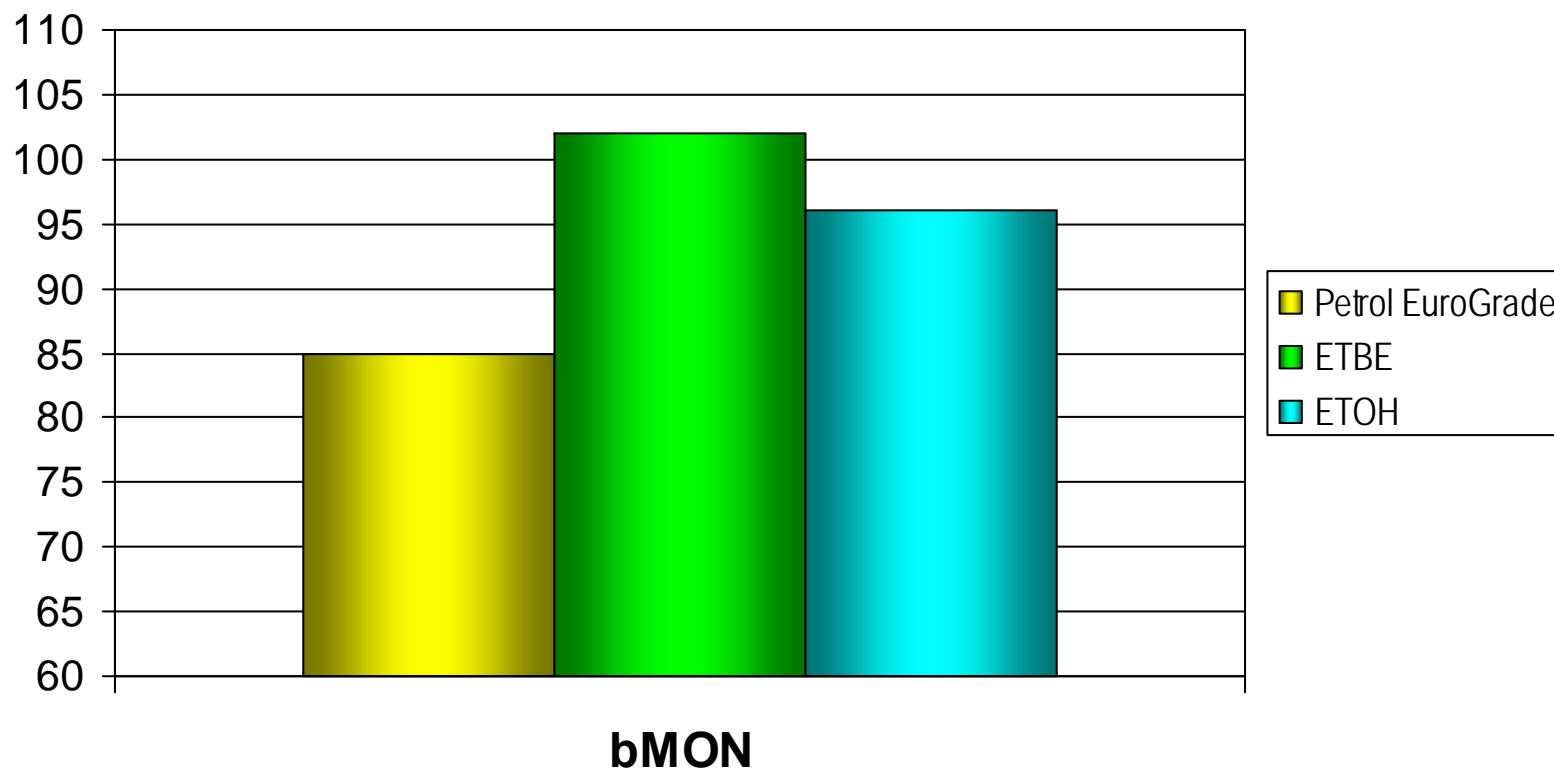


Source: CONCAWE's "RUFIT" report N° 6/78 (Dec. 1978)

ETBE Delivers **Superior MON** Contribution: Most EU Refineries “*MON-Limited*”



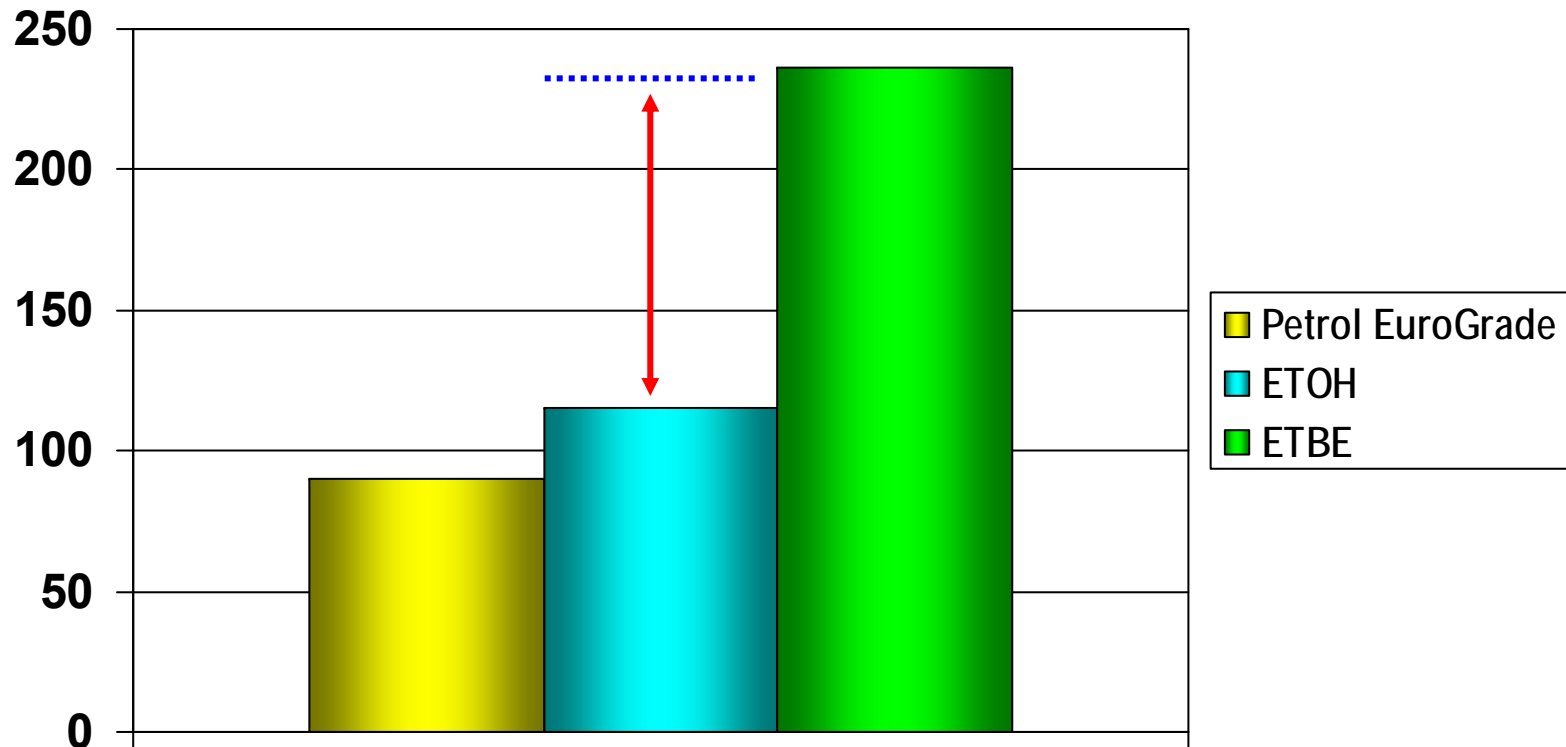
MON = %v/v Equiv. 2,2,4-trimethylpentane in i-C8/n-C7



ETBE Delivers > Twice “Octane Barrel” @ same Ethanol-equivalent



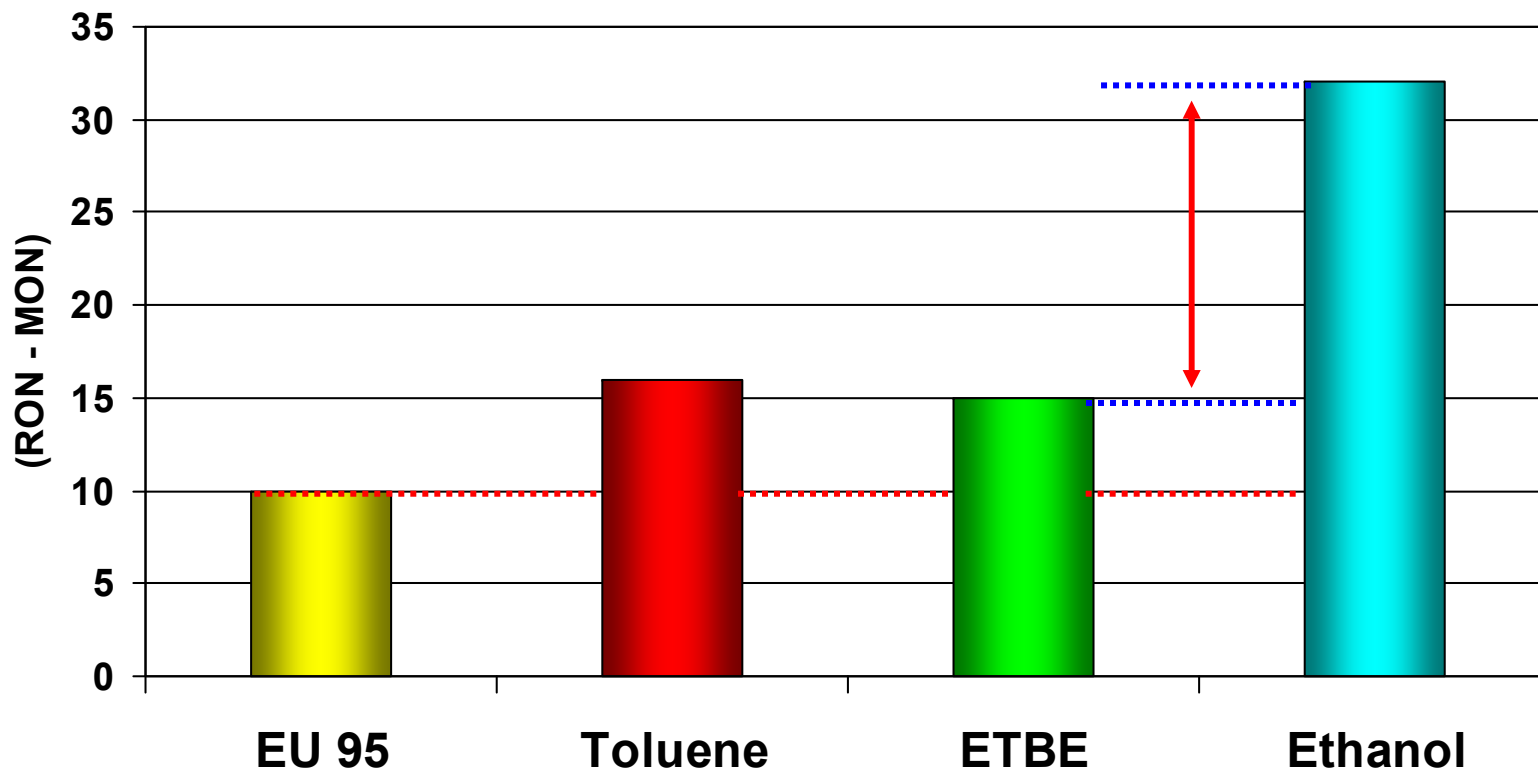
Octane “Barrell” = $(MON+RON)/2$ “Volume”
@ same ETOH-Equivalent



ETBE Helping Refiners vs. Octane Sensitivity Balance



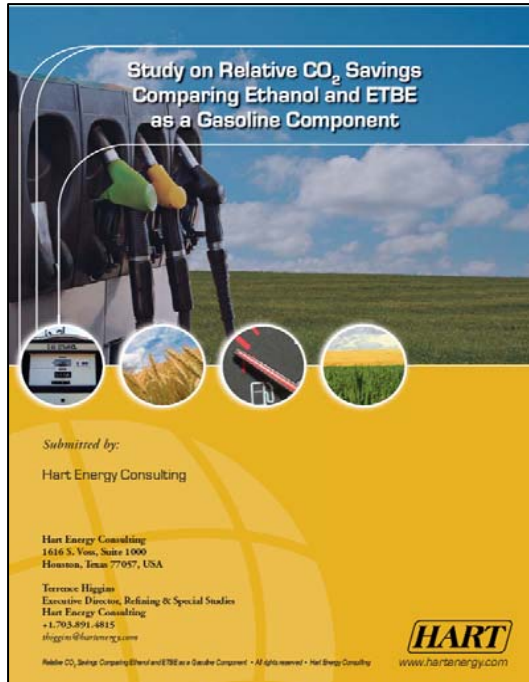
Octane Numbers = %v/v Equiv. 2,2,4-trimethylpentane in i-C8/h-C7



ETBE Reduces CO₂ Emissions

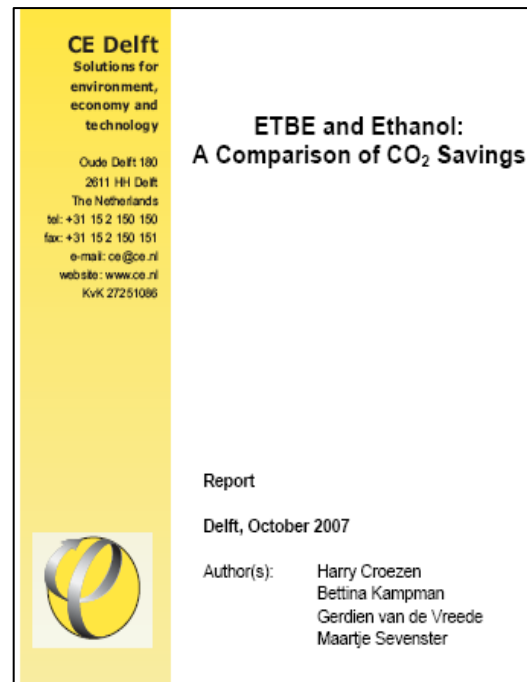


HART July 2007



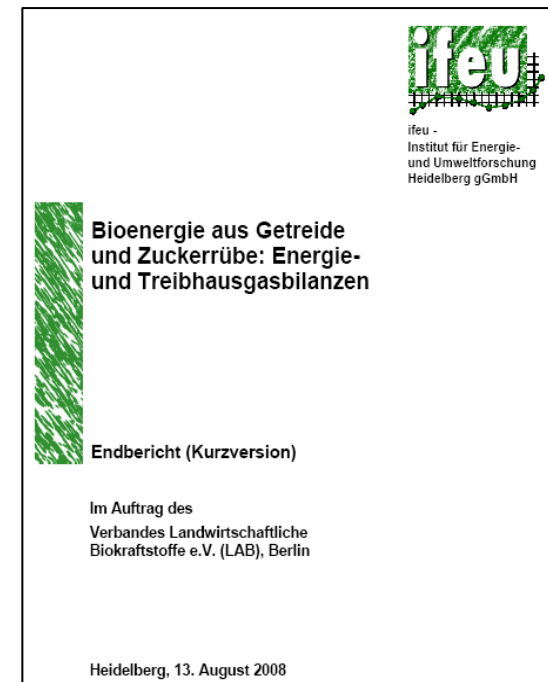
“The use of bio-ETBE reduces refining crude-oil need and processing intensity, requires less fuel and, implying relevant petrol composition changes, allows the reduction of carbon factor and lesser CO₂ emissions”

CE-Delft October 2007



“This study indicated that, when bio-ETBE is used, the resulting modification of refinery operations determine a significant reduction of greenhouse gases emissions”

IFEU August 2008



*“Best results by far are obtained when ethanol is converted to bio-ETBE.
The use of ETBE can allow the saving of 4 times the primary energy required to produce its fossil alternative.
IFEU recommends to exploit the whole potential of bio-ETBE”*

ETBE CO₂ Performance Studies: **References**



Study on Relative CO₂ Savings Comparing Ethanol and ETBE as a Gasoline Component
Hart Energy Consulting
July 2007



ETBE and Ethanol: A Comparison of CO₂ Savings
CE-Delft
October 2007

http://www.ce.nl/publicatie/etbe_and_ethanol%3A_a_comparison_of_co2_savings/715?PHPSESSID=37ad2bd9915bcf5711aed6292578b595



Bioenergy from grain and sugar beet: Energy and greenhouse gas balances
IFEU - Institute for Energy and Environmental Research Heidelberg
August 2008
<http://papers.sae.org/2009-01-1951>



Ethyl Tertiary Butyl Ether - A Review of the Technical Literature
SAE
June 2009
<http://papers.sae.org/2009-01-1951>

The impact of ethanol and ETBE blending on refinery operations and GHG-emissions
ELSEVIER - Energy Policy
2009

[http://www.ce.nl/art/uploads/file/Artikelen%20\(medewerkers\)/EnergyPolicy_TheImpactofEthanolandETBEblending_HCBKa.pdf?PHPSESSID=4d91cd6d759b670b5c0f4d0c98735687](http://www.ce.nl/art/uploads/file/Artikelen%20(medewerkers)/EnergyPolicy_TheImpactofEthanolandETBEblending_HCBKa.pdf?PHPSESSID=4d91cd6d759b670b5c0f4d0c98735687)

GHGs Emissions Calculation **Formula:** A Deeper Look

$$\frac{-2 \pm \sqrt{2^2 - 4(1)(1)}}{2(1)}$$

$$\frac{-2 \pm \sqrt{4-4}}{2}$$

$$\frac{-2 \pm \sqrt{0}}{2} \rightarrow \frac{-2}{2} \rightarrow -1$$

$$E = e_{ec} + e_l + e_p + e_{td} + e_u - e_{sca} - e_{ccs} - e_{ccr} - e_{ee}$$

where

E = total emissions from the use of the fuel;

e_{ec} = emissions from the extraction or cultivation of raw materials;

e_l = annualised emissions from carbon stock changes caused by land use change;

e_p = emissions from processing;



e_{td} = emissions from transport and distribution;

e_u = emissions from the fuel in use;



e_{sca} = emission savings from soil carbon accumulation via improved agricultural management;

e_{ccs} = emission savings from carbon capture and geological storage;

e_{ccr} = emission savings from carbon capture and replacement; and

e_{ee} = emission savings from excess electricity from cogeneration.





Bio-Ethers allow Energy Processing Saving during Refinery Fuel Formulation !



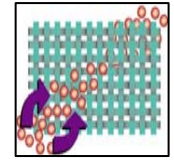
$$E = e_{ec} + e_l + \mathbf{e_p} + e_{td} + e_u - e_{sca} - e_{ccs} - e_{ccr} - e_{ee}$$

Emissions from **Processing**

$$\mathbf{e_p} = \text{process energy} \left\{ \begin{array}{l} + \text{consumed "upstream"} \\ - \text{saved "downstream"} \end{array} \right.$$

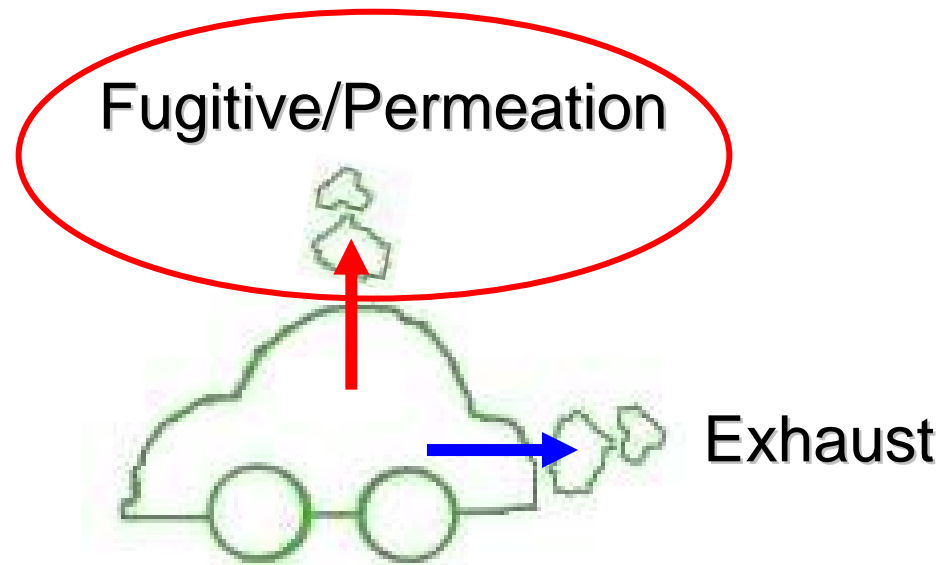


Converting ETOH into **Ethanol-Ethers** **Avoids** Vehicle **Permeation Emissions** & Secondary **CO₂ Formation**



$$E = e_{ec} + e_l + e_p + e_{td} + \mathbf{e_u} - e_{sca} - e_{ccs} - e_{ccr} - e_{ee}$$

Emissions from the **Fuel in Use**





ETBE: The MTBE “Substitution” Effect



$$E = e_{ec} + e_l + e_p + e_{td} + e_u - e_{sca} - e_{ccs} - e_{ccr} - e_{ee}$$

Emissions from ... land **use change**

LUC = Land Use Change



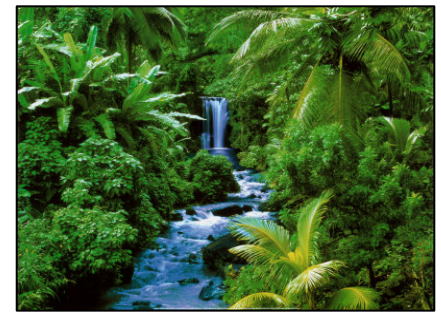
ARID LANDS

GOOD
→



ENERGY CROPS

←
BAD

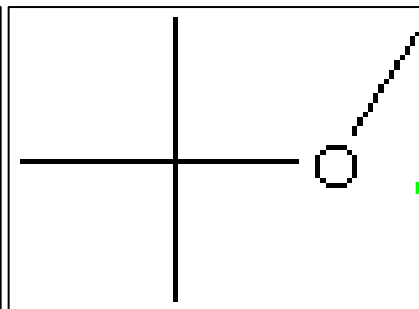


RAINFOREST

PUC = Plant Use Change



FOSSIL-ORIGIN

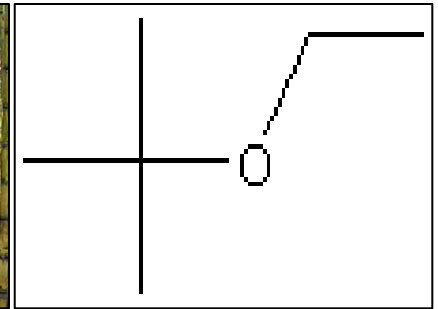


MTBE

GOOD
→



BIO-ORIGIN



ETBE

Conclusion

*Ethanol-Ethers, via their Multiple
Valuable Technical and Environmental
Features, Contribute to Enhance Fuel-
ETOH Sustainability Pathway, including
very Relevant CO₂ Emissions Reduction
Improvement*